STUDIES ON THE IMIDAZOLE SERIES

XLVI. Alkyl-, Aryl-, and Acyl-Substituted Imidazo[2,1-b]thiazoles*

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The reaction of 2-mercaptoimidazole and its 4(5)-aryl- and 4,5-diaryl-substituted derivatives with α -halogenoketones has yielded a number of alkyl-, acyl-, and aryl-substituted imidazo[2,1-b]thiazoles.

3-Alkyl-, 3-aryl-, 2,3-diaryl-, 3,5,6-triaryl-, 3-alkyl-5,6-diaryl-, 2-acyl-3-alkyl-, and 2-acyl-3-alkyl-5,6-diaryl-imidazo[2,1-b]thiazoles have not been described in the literature. In preceding investigations [2-4] we have studied the reaction of 2-mercaptoimidazole and 2-mercapto-4,5-diphenylimidazole with α-halogenoketones of the aliphatic and aliphatic-aromatic series, and also the reactions of 2-mercapto-4(5)-phenyl- and 2-mercapto-4(5)-(p-nitrophenyl)imidazoles with p-bromo- and p-nitrophenyl bromides. The performance of these reactions in ethanol in the presence of sodium ethoxide leads to the 2-(acylalkylthio)imidazoles I-XII. In the IR spectra of these compounds, unlike the IR spectra of the 2-(formylalkylthio)imidazoles [1, 5] and the 2-(acylmethylthio)imidazolines [6], which exist in the solid state as the tautomeric forms, 3-hydroxyimidazo[2,1-b]thiazoles and 3-hydroxy-5,6-dihydroimidazo[2,1-b]thiazoles, respectively, there are four bands of the stretching vibrations of the CO group in the 1675-1730 cm⁻¹ region, which shows their structure as imidazol-2-ylthiomethyl ketones.

When 2-mercaptoimidazole and 2-mercapto-4,5-diphenylimidazole are boiled with chloroacetone and with 3-chloropentane-2,4-dione in ethanol or butanol in the absence of alkali, imidazo[2,1-b]thiazole derivatives (XIII, XVII, XXI, XXII) are formed. The reaction of 2-mercaptoimidazole with aliphatic-aromatic halogenoketones (phenacyl bromide and desyl chloride) under similar conditions stops at the stage of the 2-(phenacylthio)- and 2-(desylthio)imidazoles (I, IV).

The 2-(acylmethylthio)imidazoles having residues of aliphatic ketones, such as IX, readily split out a molecule of water on being boiled in ethanol or butanol in the presence of HCl, being converted into derivatives of imidazo[2,1-b]-thiazole (XXI). The imidazol-2-ylthioketones containing residues of aliphatic-aromatic ketones undergo no change under these conditions. They cyclize to form aryl-substituted imidazo[2,1-b]thiazoles only under the action of strong water-abstracting agents such as phosphorus oxychloride. In this way compounds I-IV, VI, and VII yielded XIV-XVI and XVIII-XX (Table 1).

EXPERIMENTAL

2-(Acylalkylthio)imidazoles (I-XII, table). A) To a solution of sodium ethoxide prepared from 0.01 g-atom of metallic sodium and 15-60 ml of absolute ethanol was added 0.01 mole of 2-mercaptoimidazole [7], 2-mercapto-4(5)-phenylimidazole [8], 2-mercapto-4(5)-(p-nitrophenyl)imidazole [8], or 2-mercapto-4,5-diphenylimidazole [9] and 0.01-0.0105 mole of an α -halogenoketone (chloroacetone, 3-chloropentane-2,4-dione, desyl chloride; and in all the other examples the corresponding bromoketones). The mixture was stirred at 60-65°C for 1-2 hr and was then boiled for 5-10 min (until the solution was neutral) and cooled, and the precipitate was filtered off and washed with water. The evaporation of the ethanolic mother solution to small volume gave an additional amount of product. Compounds I-V were isolated after dilution of the reaction mixture with water.

B) A solution of 0.03 mole of 2-mercaptoimidazole and 0.03 mole of phenacyl bromide in 30 ml of ethanol was boiled for 1 hr, poured into water, and neutralized with sodium bicarbonate, and the precipitate was filtered off. Yield of I, 5.95 g (91%). Compound IV was obtained similarly, yield 89%.

^{*}For part XLV, see [1].

4.4	%	Yield,	98	66	8			8 6			97	200	5 6	25				22.0	<u>5</u>			
1. 2-(Acylalkylthio) imidazolines ^a $\frac{R^3 - 1}{1 - x_{11}}$ and Imidazo[2,1-b]thiazole Derivatives ^{a $R_3 - 1 - x_{11}$} $\frac{R_3 - 1}{1 - x_{11}}$	Ī	S	14.69	(2.18	10.89	8,59	9,45	7.72	10.40	8,66	7.14	7	27.14	21.49	11.49	13.07	34.80	11.23	98	80.0	9.81	9.65
	Calculated, %	z	9.43 10.79	15,96 12.18	9.52 10.89		12.38	0.05	9.09	7.56	6.23	10.11	11.86 27.14	9.39	10.04	17.13.13.07			13.08	10.50	8.57	8.43
		halogen	26.89	1	1	21,41	;	19.10		1	17.78	i	1		28.63	I	16.36	1	1	i	10.85	ï
		H	4.62		4.79	3,51	3,86	2,89	5.23	4.90	3.81		3.41	3.38	2.53		4,19	4.59	3.45		4.63	
		U	60,53	15.09 11.98 50.18	11,07 69.36	54.70	60.16	7.32 48.81	70,10	4.57	31.47	66,49	30,50	44.28	47.32	53,87	44.34	71.54	13,13 10 03 63.54	51.01	9.88 66.14	72,26
	Found, %	v	3,03 14,46 60,53 9,31 10,24 44,46	11.98	11,07	8.50	9,61	7.32	10.51	8.87	6.15 6.94	7.91	27.05	9.28 21.34	99.1	12,74	15.21	1.04	10 03	7,88	9.88	9.44
		z	13,03	15.09	9.54	7.42	12,18	10.26	9.15	7.12	6.15	10.22	11.47	9.28	10.04	16,56	13.09	9.29	13,13	9.78	8.61	8.35
		рзјовеп	- 26	1	1	21.13	1	19.10	!	1	18.23		1	1	28,70		16.22	1	l		11.00	1
		н	4.95 9.84	3.50	4.60	3.52	3.90	3,05	535	2.00	4.07	4.29	3,66	3,24	2.56	3,09	4.22	4.30	3,42	2,62	4.57	4.96
		Ú	60,32	50.33	69.17	54.99	59.89	49.02	70.53	74.60	09.19	66.42	30,35	44.63	47,30	54.17	43,99	71,82	63,94	51,60	66.54	72.69
	Empirical formula		C.H.10N2OS C.H.BrN.OSb	Cu Hanao.S	C17H14N2OS	C ₁₇ H ₁₃ BrN ₂ OS ^c	C ₁₇ H ₁₃ N ₃ O ₃ S	C ₁₇ H ₁₂ BrN ₃ O ₃ S	O. C. H. O. O. O.	C ₂₃ H ₁₈ N ₂ OS	C20H17BrN2OS	C23H17N3O3S	$C_6H_6N_2S \cdot H_2SO_4^e$	C11H8N2S · H2SO4	C11H7BnN2S	C11H7N3O2S	C.H.N.OS · HCI 8	C1,H1,N,S · 1/,H,O	C.H., N.O.S	CirHinBrN O.S	ClaHiAN,S. HCI	C ₂₀ H ₁₆ N ₂ OS
		Mp, °C	162—163,5	157—158		152-153	- 1	177—179	147-148 5	180—181	185-186	186-188	134-135	221 - 223	125,5—127	214 - 215	>240decomp.	153 154	204-206	261-262	243—245	190—191
	<u>~</u>		II	:1	: #	CH	$C_6H_4NO_{2}-p$	CeH4NO2-p	C6114102-6	i i	i i	Î.H.	Ę	Ή	Ħ	I	Ξ	Ξ	C.H.NO0	C.H.NOn	CH	C.H.
	°24		ΞJ	Į	ΞΞ	Ξ	Ξ	Į:	u H	i i	i i	C,H	Ţ	Ξ	Η	Ξ	Ξ	Ξ	;;	Ξ,	C.H.	$C_{\rm eH}^2$
		R ²	C,Hs,	CHINO.	C.H.	C.H.Br-p	C,H,	CeH4Br-p	֓֞֞֞֟֓֓֟֓֟֓֓֓֓֟֟֓֓֓֟֟֓֓֟֟֓֓֟֟֓֓֟֟֓֓֟֟֓֓	1 1 1 1	C.H.Rr-n	C.H.NOp	CH3	C,H,	C.H.Br-p	C.H.NOb	CH	ji T	i i	C.H.Br-n	CH.	CH3
		-~	H	= =	η, Η,	Ī	H	H.	֖֭֓֞֞֟֞֟֝֟֟֟֝֟֟֓֓֓֓֓֟֟֓֓֓֓֓֓֟֟֓֓֓֓֓֟֟֓֓֓֓֟֓֓֓֟֜֟		ΞΞ	Ξ	Ξ	Ξ	Ξ	Ξ	CH.CO	, i	, I		:=	CH3CO
Table 1	pu	Compou		111	<u> </u>	>	Ν	III	III A	< ×	~ :×	XIIX	XIIIa	XIVa	X	IXX	XVIIa	XVIII	XIX	XX	XXIA	IIXX

XIII, XIV, XVII, and XXI are oily liquids. ^bPicrate with mp 172-173°C (from 30% ethanol). ^cPicrate with mp 167-169°C (from ethanol). ^dPicrate with mp 173-175°C (from 30% ethanol). ^ePicrate of the base XIII with mp 161-164°C (from ethanol). ^EValues of ν CO in the IR spectrum (UR-10, inparaffin oil), cm⁻¹: I) 1681; II) 1680; IV) 1678; V) 1675; VI) 1676; VII 1692; II and XIII from 30% ethanol; III, V, XIVa, XVIIa, XXIa, and XXII from ethanol; VI and VIII-XII from butanol; VII from dichloroethane; XIX from ethanol-dichloroethane (1:2); XX from dichloroethane-dioxane (2:1); XIV from butanol-dimethylformamide (10;1); and XIII by precipitation with ether from ethanol. The bases aFor analysis, the substances were purified by crystallization: I, IV, IX, XV, and XVIII from 50% ethanol; VIII) 1678; IX) 1728; X) 1689; XI) 1680; XII) 1689; XIIIa) 1648. Imidazo[2,1-b]thiazole derivatives (XIII-XXII). A) A solution of 0.02 mole of a 2-mercaptoimidazole derivative and 0.021 mole of an α -chloroketone in 10-50 ml of butanol was boiled for 2-3 hr and cooled, and the precipitate of the hydrochloride (XIIIa, XVIIa, XXIIa) was filtered off and washed with ether. Evaporation of the mother liquor yielded additional amounts of these substances. The decomposition of the hydrochlorides with sodium bicarbonate in aqueous solution yielded the bases XIII, XVII, XXI, and XXII. The base XVII was also obtained by performing the reaction in ethanol (boiling for 3 hr).

B) To 0.01 mole of IX in 20 ml of butanol was added 5 ml of 10% ethanolic hydrogen chloride, the mixture was boiled for 3 hr and cooled, and the precipitate was filtered off and washed with ether. The yield of XXIa was 2.56 g (78.5%).

C) A mixture of 0.005 mole of a 2-(acylalkylthio)imidazole (I-IV, VI, VII) and 10-15 ml of POCl₃ was boiled for 7 hr (in the preparation of XIV, XVIII, and XIX), 14 hr (XVI), or 17 hr (XV, XX) and then the POCl₃ was distilled off in vacuum, the residue was treated with 10-20 ml of cold water and made alkaline with sodium carbonate or ammonia, and the precipitate was filtered off and washed with water. After the neutralization of the solution, XIV and XVIII were extracted with chloroform.

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